BUBNOV, N.N.; KRASNOVSKIY, A.A.; UMRIKHINA, A.V.; TSEPALOV, V.E.; SHLYAPINOTOKH, V.Ya.

Electron paramagnetic resonance spectra observable during the illimination of plant leaves and photoreduction of chlorophyll and its analogues. Biofizika 5 no. 2:122-126 '60. (MIRA 14:4)

1. Institut khimicheskoy fiziki AN SSSR i Institut biokhimii im. A.N. Bakha AN SSSR, Moskva.

(CHLOROPHYLL)
(PARAMAGNETIC RESONANCE AND RELAXATION)

KRASNOVSKIY, A.A.; BYSTROVA, M.I.

Fluorescence spectrophotometric study of chlorophyll synthesis in homogenates of etiolated leaves. Biokhimia 25 no.1:168-179 Ja-F *60. (MIRA 13:6)

1. Institut biokhimii imeni A.N. Bakha Akademii nauk SSSR, Moskva. (CHLOROPHYLL metab.)

26623

27/1100

2/011/01/018/001/002/014

E112/E453

AUTHORS:

Krasnovskiy, A.A. and Drozdova, N.N.

TITLE:

Effect of carotene on the photochemical properties of

chlorophyl

PERIODICAL: Chemie a chemická technologie, 1961, Vol.18, No.1, p.18. abstract Ch 61-248 (Biokhimiya, 1960, Vol.25, No.2,

pp.288-295)

Carotene is without effect on the photoreduction and photo-TEXT: sensitization. It inhibits the photochemical reaction of chlorophyl, the photoreduction, photoexidation and photosensitization in redox processes. The inhibitory function is only effective if both chlorophyl and carotene are present in the same phase. 7 diagrams, 11 literature references.

Abstractor's note: Complete translation;

Card 1/1

\$/074/60/029/06/02/005

B022/B003

5.4500A

Krasnovskiy, A. A.

TITLE:

Reaction of the Reversible Photochemical Reduction of

Chlorophyll. Its Analogs and Derivatives

PERIODICAL:

Uspekhi khimii, 1960, Vol. 29, No. 6, pp. 736-759

TEXT: The investigation of the reaction of plant pigments under the action of light is one of the ways to recognize the photochemical stage of photosynthesis. In 1948 the author (Ref. 1) proved that chlorophyll is capable of a reversible photochemical reduction. The properties of the chlorophyll, its analogs and derivatives were investigated at the laboratoriya fotobiokhimii (Laboratory of Photobiochemistry) of the Institut biokhimii AN SSSR (Institute of Biochemistry of the AS USSR); thereat the significance of the individual groupings of the pigment molecules and the nature of the central metallic atom along with the photochemical properties were studied. The following investigations were made: The chlorophylls a and b of green plants, the bacteriochlorophyll of photosynthesizing bacteria, the protochlorophyll of etiolated leaves

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Reaction of the Reversible Photochemical Reduction of Chlorophyll, Its Analogs and Derivatives 80968 \$/074/60/029/06/02/005 B022/B003

and seed-cases of Cucurbitaceae, the corresponding pheophytins, various porphyrins, phycobilins and phthalocyanines which are the synthetic analogs of these pigments. Thus, methods for the separation and purification of the plant pigments indicated were elaborated and used. In investigations spectrophotometry was frequently applied in vacuum tubes adapted for measurements in Beckmann devices of the type CQ-4 (SF-4). In order to study the reactions, the fluorescence spectra, the photopotential of the inert electrode and photoconductivity in the reaction solution were measured in a series of experiments. The following methods were applied in order to investigate the active reaction intermediates and to distinguish the individual stages: 1) application of media in which the reaction intermediates are stabilized (organic compounds), 2) treatment at low temperatures, and 3) use of hydrogen donors labeled with deuterium. First of all, the reaction of sensitive photoreduction was studied by means of spectrophotometric measurement of the following substances: chlorophyll, pheophytine, ethylchlorophyllide, protochlorophyll and its derivatives, bacteriochlorophyll, bacteriopheophytin, bacterio-

Card 2/5

Reaction of the Reversible Photochemical Reduction of Chlorophyll, Its Analogs and Derivatives

80968 S/074/60/029/06/02/005 B022/B003

viridine, porphyrin, metallic complexes of chlorophyll and its analogs, phthalocyanine, carotinoids, and phycobilins. Fig. 1 shows the absorption spectra of chlorophyll and its analogs in photoreduction by ascorbic acid in pyridine, and Fig. 2 represents the absorption spectra of photoreduced forms of hematoporphyrine in the main acid media. The structure of the pigment and its ability to photoreduction are discussed in Chapter II; Chapter III compares the properties of transitional forms produced in photoreduction and reduction by means of zinc in the dark. The variation of fluorescence in photoreduction is dealt with in Chapter IV, and the variation of the redox potential $\mathbf{E}_{\mathbf{0}}$ in photoreactions is discussed in Chapter V. The molecules of the reducing substance reacting with chlorophyll and its analogs are described in Chapter VI, whereas the influence exerted by the pH of the medium is dealt with in Chapter VII. Further, descriptions are given of the formation of free radicals, of chemiluminescence in the reversible oxidation of photoreduced products of chlorophyll, of the formation of associates prior to the reaction, and of the mechanism of photoreduction. The paramagnetic electron resonance

Card 3/5

Reaction of the Reversible Photochemical Reduction of Chlorophyll, Its Analogs and Derivatives

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spectrum forming on exposure to light of the system pigment - ascorbic acid is illustrated in Fig. 3. The chemiluminescence in the oxidation of the photoreduced form of chlorophyll is indicated in Fig. 4. The scheme of the photoreduction of chlorophyll by ascorbic acid is shown in Fig. 5. The most important experimental results confirming the indicated reaction mechanism are listed in a table. Chapter II contains a description of the redox reactions sensitized by chlorophyll, and Chapter III describes photoreactions of colloidal chlorophyll solutions. The photochemical properties of chlorophyll in chloroplast globules are described in Chapter IV; reversible photochemical conversions of chlorophyll in living plant leaves in Chapter V; and Chapter VI deals with the reaction of reversible photoreduction as well as the possible mechanism of the participation of chlorophyll and its analogs in photobiological processes. Mention is made of: V. B. Yevstigneyev, V. A. Gavrilova, N. V. Vostrilova, V. I. Dulova, B. Ya. Dain, M. S. Ashkinazi, K. A. Timiryazev, V. F. Gachkovskiy, Yu. Ye. Yerokhin, G. P. Brin, A. V. Umrikhina, A. N. Terenin, O. P. Osipova, I. Timofeyeva, and F. F. Litvin. The author also refers to investigations

Card 4/5

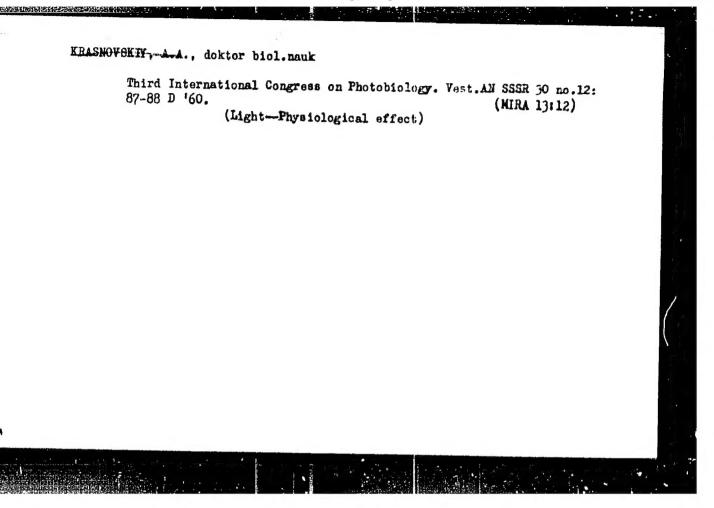
Reaction of the Reversible Photochemical Reduction of Chlorophyll, Its Analogs and Derivatives

S/074/60/029/06/02/005 B022/B003

made at the Institut khimicheskoy fiziki AN SSSR (Institute of Chemical Physics of the AS USSR). There are 5 figures, 1 table, and 112 references: 69 Soviet, 35 English, 4 German, 1 Swiss, 1 Indian, 1 Polish, and 1

ASSOCIATION: In-t tiokhimii im. A. N. Bakha AN SSSR (Institute of Biochemistry imeni A. N. Bakh of the AS USSR)

Card 5/5



S/053/60/071/01/07/011 B006/B011

AUTHORS:

Litvin, F. F., Vladimirov, Yu. A., Krasnovskiy, A. A.

TITLE:

Chemoluminescence of Chlorophylls in Photochemical Reactions

PERIODICAL: Uspekhi fizicheskikh nauk, 1960, Vol. 71, No. 1, pp. 149-156

TEXT: The authors investigated the red chemoluminescence in the oxidation of a reduced form of chlorophyll and the chemoluminescence in the irradiation of chlorophyll solutions in the air, as well as the afterglow and the thermoluminescence of chloroplast films, extracts and pure pigments, and the dependence of these effects on temperature and the presence of oxygen. A report is made on these investigations and on the results obtained. Fig. 1 shows the extinction curves of the afterglow in green leaves and the chemoluminescence in the oxidation of a photoreduced form of the chlorophyll (in pyridine solution + ascorbic acid). The diagram contains two more curves reproducing chemoluminescence with different ascorbic acid contents in the solution. A strong dependence of the luminescence yield on the ascorbic acid content was found. Chemoluminescence was found to be a consequence of the oxidation of photo-

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Chemoluminescence of Chlorophylls in Photochemical Reactions

S/053/60/071/01/07/011 B006/B011

reduced chlorophyll; it could be only observed on pure chlorophyll a. Fig. 2 depicts the dependence of chemoluminescence on the oxygen and ammonia content in the solution, and Fig. 3 illustrates the kinetics of chemoluminescence of the rate in the change of the optical density in the oxidation of a photoreduced form of chlorophyll. It was further found that the luminescence quantum is emitted from the chlorophyll molecule, but that intensities are so low that no spectrum could be taken. Chemoluminescence was also observed in such cases where photoreduction of pheophytin a occured at -40°C. The authors further investigated pyridine solutions of mixtures of chlorophyll a and b, pure chlorophyll a and pheophytin a. Fig. 4 shows extinction curves of the chemoluminescence of chlorophyll solutions in pyridine after irradiation in vacuum and in the air. Results are then discussed, as were obtained from the investigation of chemoluminescence of chlorophyll in films of different material and in chloroplasts. Table 1 shows the duration of afterglow and the duration of the thermoluminescence of different films that were partly irradiated in vacuum and partly in oxygen. Fig. 5 shows extinction curves of afterglows and thermoluminescence. In general, thermoluminescence and afterglows were only observed in such cases, when there was oxygen present during the

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Chemoluminescence of Chlorophylls in Photochemical Reactions

S/053/60/071/01/07/011 B006/B011

excitation irradiation. Nor did there occur any thermoluminescence, when chloroplast films had been irradiated at nitrogen temperatures; an intensive afterglow of nonactivated oxygen occurred in this case. The longer afterglow of photosynthesized cells and tissues, chloroplast suspensions and pigment solutions, as well as the thermoluminescence of chloroplasts are assumed to have a common nature, viz. they are assumed to be caused by the chemoluminescence of photoproducts. A. N. Terenin is mentioned. There are 5 figures, 1 table, and 13 references, 5 of which are Soviet.

Card 3/3

"APPROVED FOR RELEASE: Monday, July 31, 2000 CIA-RDP86-00513R000826210

. KRASNOVSKIY, A.A.; YEROKHIN, Yu.Ye.; FEDOROVICH, I.B.

Fluorescence of green photosynthesizing bacteria and the state of bacteriovirdin in them. Dokl. AN SSSR 134 no.5:1232-1235 0 '60. (MIRA 13:10)

1. Institut biokhimii im. A.N.Bakha Akademii nauk SSSR i Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova. Predstavleno akademikom A. N. Tereninym.

(CHLOROBIUM)

(FLUORESCENCE)

(VIRIDIA)

5.4500

1206, 1273, 1153, also 3612

\$/020/60/135/005/042/043

F016/B052

AUTHORS:

Krasnovskiy, A. A. and Pakshina, Ye. V.

TITLE:

Reversible Photoreduction of Bacteriochlorophyll and Its Participation in Processes of Photochemical Electron Trans-

fer

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol. 135, No. 5,

pp. 1258-1261

TEXT: The authors describe the formation of active photoproducts in the photoreduction of bacteriochlorophyll (from Rhodopseudomonas palustris) and the interaction between these products and electron acceptor molecules taking part in the biochemical electron transfer of photosynthesis. Bacteriochlorophyll was submitted to a final chromatographic purification (Ref. 4). The photoreactions in pyridine and water were conducted in special Tunberg tubes in which measurements with Beckmann and $C\Phi-4$ (SF-4) spectrophotometers were possible. The electron donor typical of bacterial photosynthesis of Na₂S was used. The solution was evacuated by shaking for 30-60 sec and exposed to light near the infrared region. As a result, Card 1/4

Reversible Photoreduction of Bacteriochlorophyll and Its Participation in Processes of Photochemical Electron Transfer S/020/63/135/005/042/043 B016/B052

the absorption maximum of bacteriochlorophyll at 780 mm dropped rapidly, while the maximum of the photoreduced form appeared at 660 mm (Fig. 1). In the dark, this reaction became automatically reverse, and at room temperature bacteriochlorophyll is almost completely regenerated within a few minutes. Regeneration sets in immediately on exposure to air (Fig. 2). The authors believe that an extraordinarily active photoreduced form of pigment and a photoexidized form of electron donor are formed, probably in accordance with the elementary process of $X + SH \longrightarrow X + SH (X -$

molecule of bacteriochlorophyll). They also studied the potential of a platinum electrode during exposure to light of the following types of pigment in pyridine with an addition of Na₂S: chlorophyll a, bacterio-

champenfull, and bacterioviridine. The potentials of the three pigments were very similar (Fig. 3): During exposure to light, the potential changes by approximately 0.1 v in the negative direction. In the dark, the potential returned to its original value within 1 - 2 minutes. The authors assume that all three pigments form electron-active photoreduced products of the same type. In previous papers (Ref. 7) it was found that photoreduced pigments do not accumulate in the ternary system (electron donor - Card 2/4

Reversible Photoreduction of Bacteriochlorophyll and Its Participation in Processes of Photochemical Electron Transfer 3/020/60/135/005/042/043 3016/8052

chlorophyll - electron acceptor), since they rapidly react with electronacceptor molecules. However, when the reduction of these molecules is "sensitized", the photoreduced form of the sensitized pigment starts to accumulate. Hence, the authors conclude that the absence of visible photochemical changes of a sensitized pigment in the ternary system indicates a rapid reaction of the electron-acceptor molecule with the photoreduced form of pigment. Using this comparatively sensitive method the authors tested the action of the following co-factors of photosynthetic phosphorylation: Menadione, phenazine metasulfate, and riboflavin (flavin mononucleotide). Hence, they concluded that all of the three last-mentioned substances can receive electrons from the photoreduced form of chlorophyll. These experiments on similating systems prove that the above co-factors take part in the reactions of electron transfer if they are sensitized by a system of pigments (Ref. 9). The authors recommend the explanation of such elementary processes with the help of living photosynthesizing bacteria in which bacteriochlomophyll mainly appears in aggregated and ordered forms (Ref. 10). There are 4 figures and 10 references: 8 Soviet and 2 US.

Card 3/4

APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000826210(

Reversible Photoreduction of Bacteriochloro-5/020/60/135/005/042/043 phyll and Its Participation in Processes . B016/B052 of Photochemical Electron Transfer

ASSOCIATION:

Institut biokhimii im. A. N. Bakha Akademii nauk SSSR

(Institute of Biochemistry imeni A. N. Bakh of the Academy

of Sciences USSR)

PRESENTED:

July 6, 1960, by A. N. Terenin, Academician

SUBMITTED:

May 9, 1960

Card 4/4

17.1156

5.4500

1206,1273 00/2

\$/020/60/135/006/034/037

B016/B060

AUTHORS:

Litvin, F. F., Krasnovskiy, A. A., and Rikhireva, G. T.

TITLE

Luminescence of Different Forms of Chlorophyll in Plant

Loaves

PERIODICAL

Doklady Akademii nauk SSSR, 1960, Vol. 135, No. 6,

pp. 1528-1531

TEXT: The authors report here on their study of the fluorescence spectra of plant leaves (beans, Phaseolus), and on the changes these spectra undergo as eticlated leaves turn green and because of the effect of disintegrating agents (pyridine). Measurements were made with an earlier described apparatus (Ref. 5). Fluorescence was excited by a mercury-quartz lamp in the 380 - 580 mp. range and taken by a photomultiplier with a recording attachment. Fig. 1 shows the effect of low temperatures (-196°C) upon the leaves spectra. This sort of cooling is said to considerably increase the quantum emission of fluorescence of such chlorophyll forms as have maxima in the ranges 693 - 696 and 730 - 740 mp.

Card 1/3

APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000826210(

Luminescence of Different Forms of Chlorophyll in Plant Leaves

S/020/60/135/006/034/037 B016/B060

Fig. 2 shows curves of the effect of pyridine (10% aqueous solution) upon the spectra of leaves and homogenized mass of leaves (0.5 - 1% aqueous pyridine solution). Moreover, the effect of heating up to 100°C is shown here. Both the disintegrating agents and the high temperature disturb the native state of the pigment. In the authors opinion, the mentioned actions lead to a rapid decrease of fluorescence intensity in the longwave range (690 and 730 - 740 mm) as well as to a modification in the chlorophyll fine structure. It is therefore stated that the characteristics of low-temperature fluorescence spectra are bound to the native state of the pigment. The investigation was extended to "he formation of various chlorophyll forms in various stages of the chlorophyll formation process on the basis of fluorescence spectra of greening etiolated leaves of the sort "Triumf" during the first 24 h. In early stages of chlorophyll concentration the fluorescence spectrum at -1960C very much resembles the "ordinary" spectrum of leaves of low chlorophyll content measured at room temperature (Fig. 3). With progress; hlorophyll concentration the spectrum changes gradually; longwave maxima of fluorescence appear and develop to prevail in the end. A structure which is typical of the lowtemperature spectra of the green leaves appears at the same time. Fig. 4 Card 2/3

Luminescence of Different Forms of Chlorophyll in Plant Leaves

\$/020/60/135/006/034/037 B016/B060

shows the changes in fluorescence intensity (at -196°C) of etiolated leaves as they turn verdant. Summing up: plants exhibit several fluorescence maxima of chlorophyll: 675, 682, 690-696, and 730 m μ , which visibly belong to different chlorophyll forms. The authors base on the type of dependence of the appearance of the fluorescence spectra on temperature, on the effect of disintegrating agents, and the chlorophyll amount; to believe that the longwave maxima 683 - 696 and 730 mm belong to aggregated chlorophyll forms with a special type of molecular packing. There are 4 figures and 7 references: 4 Soviet and 3 US.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova (Moscow State University imeni M. V. Lomonosov). Institut biokhimii im. A. N. Bakha Akademii nauk SSSR (Institute of Biochemistry imeni A. N. Bakh of the Academy of Sciences USSR)

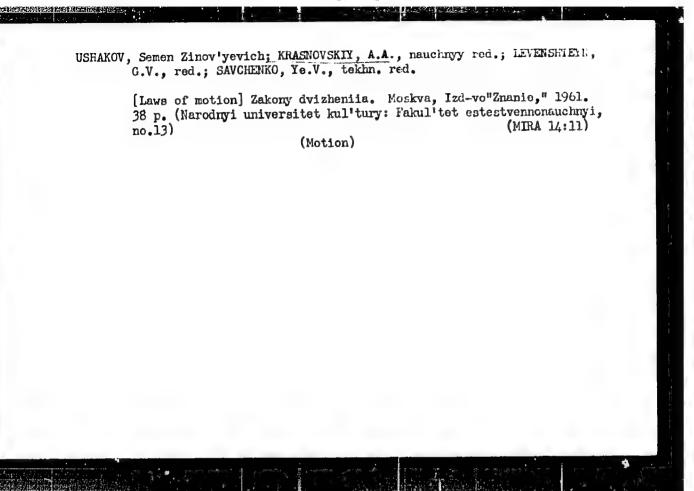
PRESENTED:

August 27, 1960, by A. N. Terenin, Academician

SUBMITTED:

August 20, 1960

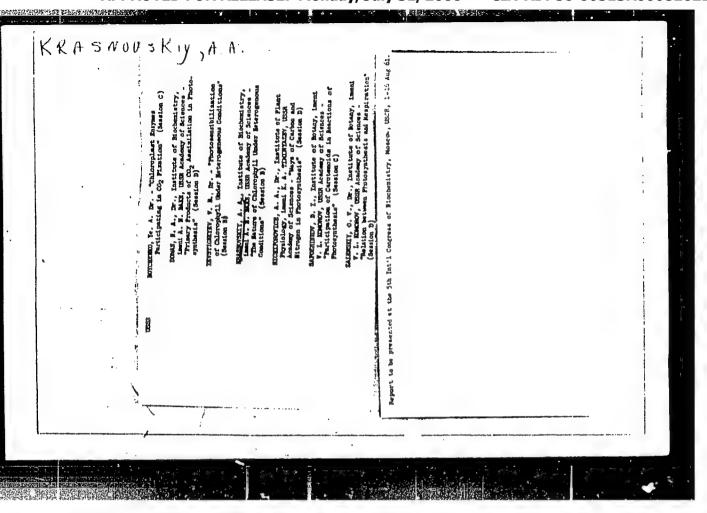
Card 3/3



"APPROVED FOR RELEASE: Monday, July 31, 2000 CIA-RDP86-00513R000826210

Studies of the State of Figurets and their traces of the Indiatal States.

report submitted for the 1st Intl. Biophysics Commerce, Sinch who, 21 July - 4 August 1961.



ZRECTOVIZIY, A. A., LITVIN, 3. S., VLIDITIROV, VI. C. (1993)

"Use of a Photon Counter in the Investination of Chicrophyll Chemilaminescence in Protobiochemical Reactions."

Report presented at the 5th International Blochemistry Congress, Moscow, 10-16 August 1961

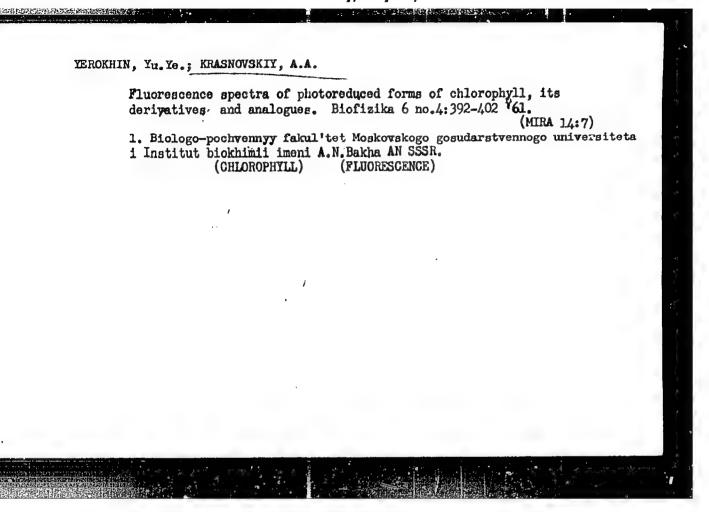
WORKERA, L. M., PAKOHI A, W. V., UMRESHIM, K. V., BREE, T. ...,

WORKERA, L. M., DROZDWA, N. N., YEROKHIM, W. W. (USSR)

"Different Forms of Chlorophyll and its Analogues and their
Role in Processes or Photochemical Electron (or lydrogen)

Report presented at the 5th International Biochemistry Congress,

Moscow, 10-16 August 1961



DOMAN, N.C.; KRASNOVSKIT, A.A.; ROMANOVA, A.K.; VOROB'YEVA, L.M.; PAKSHINA, Ye.

V.; TERENT'YEVA, Z.A.

Chlorophyll synthesis and carbon dioxide fixation in etiolated barley eseddings during exposure to light. Fiziol. rast. 8 no.1:3-12 '61.

(MIRA 14:3)

1. A.N. Bakh Institute of Biochemistry, U.S.S.R. Academy of Sciences, Moscow.

(Chlorophyll) (Photosynthesis)

"APPROVED FOR RELEASE: Monday, July 31, 2000 CIA-RDP86-00513R000826210

Study of the photoreduction of chlorophyll in the presence of electron acceptors. Biokhimiia 26 no.5:859-871 S-0 '61. (MIRA 14:12)

1. Institutes of Biochemistry, Academy of Sciences of the U.S.S.R., Noscow. (CHLOROPHYLL)

KRASNOVSKIY, A.A.; BYSTROVA, M.I.; SOROKINA, A.D.

Fractionation of different pigment forms in the homogenates of ethiolated and illuminated leaves. Dokl.AN SSSR 136 no.5:1227-1230 F **161. (MIRA 14:5)

1. Institut biokhimii im. A.N.Bakha AN SSSR, Mosk. gos. universitet im. M.V.Iomonosova. Predstavleno akad. A.N.Tereninym. (CHLOROPHYLL)

RRASHOVSEIY, A.A.; ERIN, G.P.

Photocatalytic action of ZnO and TiO₂ in the reactions involving the liberation of exygen. Dokl. AN SSSR 139 no.1:142-145 J1 '61.

(MIRA 14:7)

1. Institut biokhimii im. A.N. Bakha AN SSSR. Predstavleno akademikom A.N. Toreninym.

(Zinc oxide) (Titanium oxide) (Photochemistry)

LITVIN, F.F. Prinimali uchastiye: MALKOVA, M.G.; KOSAREVA, Ye.A.;

SISAKYAN, N.M., akademik, glav. red.; BAYEV, A.A., zam. glav.

red.; KRASNOVSKIY, A.A., red. tem; VETROVA, I.B., red. indva; DOROMHINA, I.N., tekhn. red.

[Mechanism of photosynthesis; symposium VI]Mekhanizm fotosinteza; simpozium VI. Predsedateli: Kh.Tamia (IAponiia),
A.A. Krasnovskii (SSSR). Moskva, Izd-vo Akad. nauk SSSR,
1962. 386 p. (Its: Trudy) (MIRA 16:1)

1. International Congress of Biochemistry. 5th, Moscow, 1961.
2. Chlen-korrespondent Akademii nauk SSSR (for Krasnovskiy).

(Photosynthesis—Congresses)

KRASNOVSKIY, A.A.: YEROKHIN, Yu.Ye.: KHUN YUY-TSYUN* [Hung Yu-ch'in] Fluorescence of aggregated forms of bacteriochlorophyll, bacterioviridin, and chlorophyll in connection with the state of pigments in photosynthesizing organisms. Dokl. AN SSSR (MIRA 15:3) 143 no 2:456-459 lbr 162. 1. Institut biokhimii im. A.N.Bakha AN SSSR i Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova. Predstavleno akademikom A.N.Tereninym.
(FLUORESCENCE)

PIGMENTS) (BACTERIA, AUTOTROPHIC)

KRASNOVSKIY, A.A.; DROZDOVA, N.N.

Reversible photochemical reduction of polymethine dyes. Dokl.
AN SSSR 145 no.1:129-132 J1 '62. (MIRA 15:7)

1. Institut biokhimii imeni A.N.Bakha AN SSSR. Predstavleno akademikom A.N.Tereninym.
(Dyes and dyeing) (Photochemistry)

KRASNOVSKIY, A.A.; BRIN, G.P.

Inorganic models of Hill's reaction. Dokl. AN SSSR 147 no.3:656-659 N *62. (MIRA 15:12)

1. Institut biokhimii im. A.N. Bakha AN SSSR. Predstavieno akademikom A.N. Tereninym. (Chemical models) (Photochemistry)

RUBIN, A.B.; MINCHENKOVA, L.Ye.; KRASNOVSKIY, A.A.; TUMERMAN, L.A.

Studying the average duration of the fluorescence of protochlorophyll during the greening of etiolated leaves. Biofizika 7 no.5:571-577 62. (MIRA 17:8)

l. Institut radiatsionnoy i fiziko-khimicheskoy biologii, Moskva, Biologo-pochvennyy fakulitet Moskovskogo gosudarst-vennogo universiteta imeni Lomonosova i Institut biokhimii imeni A.N. Bakha AN SSSR, Moskva.

1,5336 8/020/63/148/004/025/025 B144/B186

27 ma AUTHORS:

A., Corresponding Mamber AS USSR, Pakshina, Ye.

TITLE:

Comparative study of the formation of pheophytins from chlorophyll and its analogs in the dark and on exposure to

light

Akademiya nauk SSSR. Doklady, v. 148, no. 4, 1963, 935-938 PERTODICAL:

TEXT: Chlorophyll a and b, protochlorophyll, bacteriochlorophyll and bacterioviridin were converted to pheophytins by treating their ethereal solutions with aqueous solutions of HCl or oxalic acid. The reaction was studied spectrometrically in acetone and pyridine, both containing 10% water. The effect of light (104 erg/cm²·sec) was investigated in vacuo to prevent secondary photooxidation. At 22°C and with 10-5 mole/l pigment and 2.10-2 mole/1 oxalic acid, the time (in min) needed for the halfconversion of the Mg complex in the dark was:

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Comparative study of the forma	stion Bi	/020/63/148/004/025/025 144/B186
	acetone (both conta	pyridine aining 10% water)
bacterioviridin	0.5	25
chlorophyll a	4	240
protochlorophyll	25	>400
bacteriochlorophyll	25	> 100
	25	>400
chlorophyll b	105	>400
	differ widely. The the "half-isolated" recophytinization of topentanone ring. It is and at -20°C. Whinization of bacterimore than in the deptochlorophyll was contacted within 10-15.	ey prove to be independent "double bonds in the bacterioviridin may be The effect of light was noe of 2·10-2 - 2·10-2 mole/1 hile no effect at all was ioviridin was rapid in ark. Photopheotynization completed in pyridine 5 min and in pyridine
chlorophyll b In acctone the reaction rates of the degree of reduction of initial pigment. The rapid ph due to the absence of the cyclestudied in aqueous acctone soloxalic acid at room temperature observed in acctone, pheophytipyridine, but not accelerated of chlorophyll a and b and procontaining 2.10 ⁻² mole/l oxali	differ widely. The the "half-isolated" recophytinization of topentanone ring. It is and at -20°C. Whinization of bacterimore than in the deptochlorophyll was contacted within 10-15.	ey prove to be independent "double bonds in the bacterioviridin may be The effect of light was noe of 2·10-2 - 2·10-2 mole/1 hile no effect at all was ioviridin was rapid in ark. Photopheotynization completed in pyridine 5 min and in pyridine

S/020/63/148/004/025/025 Comparative study of the formation ... B144/B186

labile photoreduction products form with pyridine or oxalic acid owing to their electron-donor properties. To verify the expected inhibition by electron-acceptors or polyenes, tests were made with addition of methyl red, carotene and naphthacene. These compounds inhibited the photoreaction

in concentrations of 10^{-3} - 10^{-5} mole/1 independently of the acid used but did not affect pheophytinization in the dark. Owing to the vacant electron pairs at the N of the pyrrole rings the hydrogen ions are bound before Mg is driven out; but it is still unclear whether the incorporation of the protons proceeds in one or more stages. Photopheophytinization which has been observed only in the absence of stable photoreduction products is inhibited by the same substances as photoreduction and is probably the cause of reversible photoreduction. There are 2 figures and 1 table.

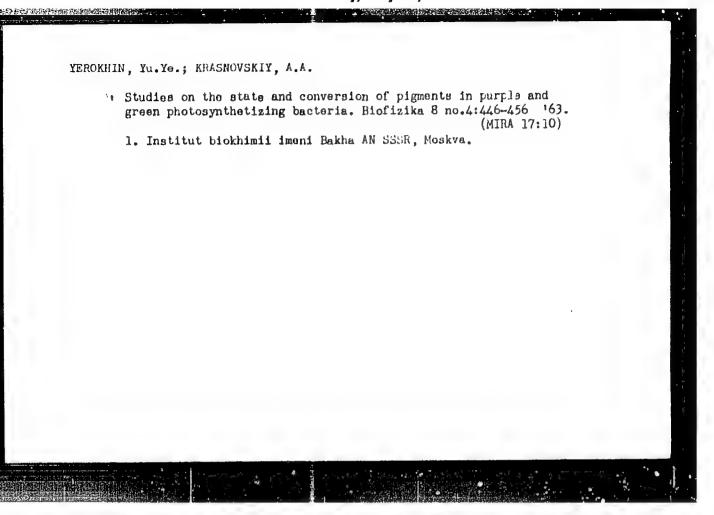
ASSOCIATION: Institut biokhimii im. A.N. Bakha Akademii nauk SSSR

(Institute of Biochemistry imeni A.N. Bakh of the Academy of

Sciences USSR)

SUBMITTED: October 15, 1962

Card 3/3



KARAPETYAN, N.V.; LITVIN, F.F.; KRASMOVSKIY, A.A.

Luminescence variations in studying the differential spectra of photosynthesizing organisms. Dokl. AN SSSR 149 no.6:1428-1431
Ap '63. (MIRA 16:7)

1. Institut biokhimii im. A.N.Bakha AN SSSR 1 Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. 2. Chlen-korr spondent AN SSSR (for Krasnovskiy).

(Photosynthesis) (Luminescence) (Spectrum analysis)

KRASNOVSKIY, A.A.; BRIN, G.P.; DROZDOVA, N.N.

Oxidation-reduction conversions of benzyl nicotinamide and pyridine nucleotides photosensitized by chlorophyll. Dokl. AN SSSR 150 no.5:1157-1160 Je '63. (MIRA 1618)

1. Institut biokhimii im. A.N.Bakha AN SSSR. 2. Chlen-korrespondent AN SSSR (for Krasnovskiy). (Nicotinamide) (Codehydrogenases) (Chlorophyll) (Oxidation-reduction reaction)

L 12843-69 ACCESSION NR: AP3003233

3/0020/63/150/006/1378/1361

AUTHOR: Krasnovskiy, A. A.; Drozdova, N. N.

44

TITLE: Reversible photochemical interaction of chlorophyll, bacteriochlorophyll, and bacterioviridine with quinone and oxygen in an alcohol-glycerine medium

SUURCE: AN SSSR. Doklady, v. 150, no. 6, 1963, 1378-1381

TOPIC TAGS: chlorophyll, bacteriochlorophyll, bacterioviridine, quinone, oxygen, aicohol-glycerine medium, reversible photo-oxidation, pigment regeneration

ABSTRACT: Continuing their studies on reversible photo-oxidation in solutions, the authors determined the changes in optical density (D) in the "red" absorption maximum of pigments subjected to alternating periods of light and darkness. The medium selected, a 1:2 mixture of alcohol and glycerine, promotes photo-oxidation. In experiments on the photochemical behavior of chlorophyll a, bacteriochlorophyll, and bacterioviridine in the presence of atmospheric oxygen, at +20C and -70C, optical density was measured 15-20 sec after illumination and in the dark. No significant reversibility effects were observed unless ascorbic acid was added.

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L 12843-63 ACCESSION NR: AP3003233

At +20C, chlorophyll a showed much less fading than bacteriochlorophyll and bacterioviridine, even though the intensity and duration of illumination were much less in the experiments with the latter pigments. Lowering the temperature to -70C did not enhance the extinction of chlorophyll fluorescence by the dissolved oxygen. In experiments on the reversible fading of these pigments in the presence of quinons (same medium and temperatures), illumination at +200 did not affect the pigments! absorption maxima, but at -70C, reversible effects were noted, with almost complete pigment regeneration after standing in the dark. No interaction between the pigments and quinone occurred in the dark, and the optical density changed only after exposure to light. For all 3 pigments, the decrease in optical density in the red absorption maxima amounted to 20-25% of the initial magnitude. In the chlorophyll a-quinone experiments at -70C, an increase in absorption at 720, 550, and 500 millimicra accompanied this reversible decrease. Solutions of light-resistant pigments in a vacuum in the presence of quinone, illuminated for 15 min, showed no change in D; such a change occurred only after exposure to air and light. The observed effects are tentatively ascribed to reversible oxidation of chlorophyll and its analogs by p-quinone. Orig. art. has: 3 figures.

APPROVED FOR RELEASE: Monday, July 31, 2000

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CIA-RDP86-00513R0008262100

KRASNOVSKIY, A.A.; YEROKHIN, Yu.Ye.; GULYAYEV, B.A.

Effect of temperature on the luminescence of bacterioviridin and its state in photosynthetizing bacteria. Dokl. AN SSSR 152 no.5:1231-1234 0 '63. (MIRA 16:12)

1. Institut biokhimii im. A.N.Bakha AN SSSR i Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. 2. Chlen-korrespondent AN SSSR (for Krasnovskiy).

KRASNOVSKIY, A.A.; BRIN, G.P. Participation of reduced pyridine nucleotides in photochemical oxidation-reduction reactions. Dokl. AN SSSR 153 no.1:212-215 N '63.

1. Institut biokhimii im. A.N. Bakha AN SSSR. 2. Chlen-korrespondent AN SSSR (for Krasnovskiy).

(MIRA 17:1)

KRASNOVSKIY, A.A.; DROZDOVA, N.N.

Study of the photochemical reactions of chlorophyll and photosensitization in viscous media. Dokl. AN SSSR 153 no.3:721-724 N 63. (MIRA 17:1)

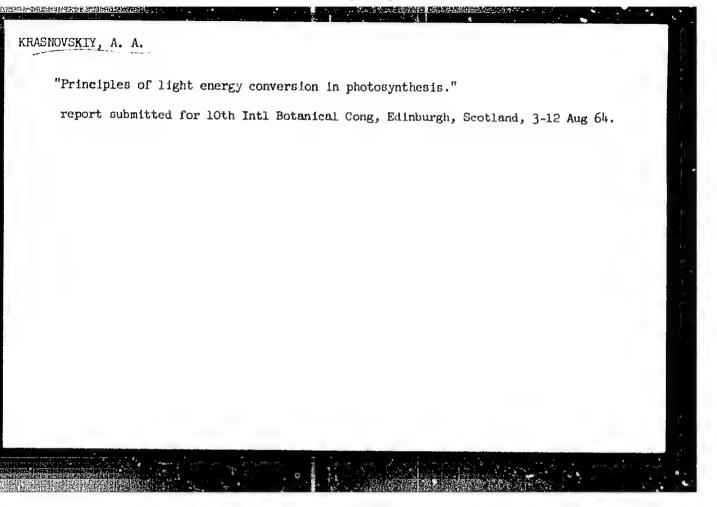
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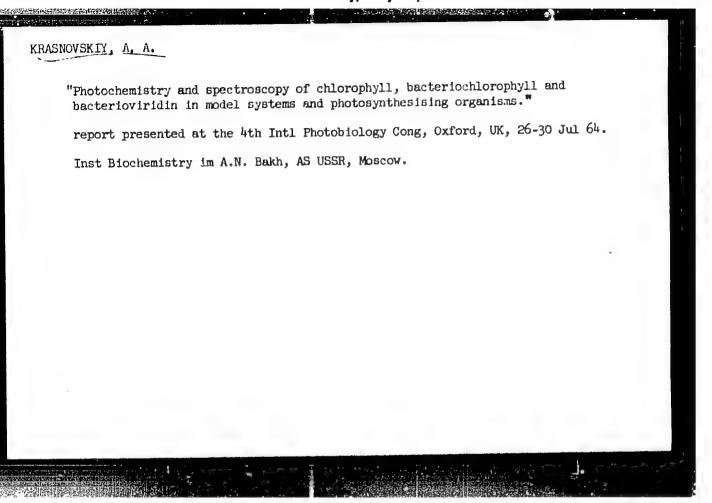
RIKHIREVA, G.T.; KRASHOVSKIY, A.A.; KAYUSHIN, L.P.

Biophysics: Relation between the state of chlorophyll and electron paramagnetic resonance spectra in plant leaves.

Dokl. AN SSSR 156 no.6:1451-1454 Je '64. (MIRA 17:8)

l, Institut biologicheskoy fiziki AN SSSR i Institut biokhimii imeni A.N. Bakha AN SSSR. 2. Chlen-korrespondent AN SSSR (for Krasnovskiy).





\$/0217/64/009/004/0423/0427

AUTHOR: Umrikhina, A. V.; Golubev, I. N.; Kayushin, L. P.; Krasnovskiy, A. A.

TITLE: A study of the paramagnetic properties of chlorophyll and its analogs

SOURCE: Biofizika, v. 9, no. 4, 1964, 423-427

TOPIC TAGS: tetrapyrrol pigment, chlorophyll, ethyl chlorophyllide, pheophytin, phthalocyanin, magnesium phthalocyanin, EPR signal, paramagnetic property, light effect, chlorophyll aggregation, EPR signal temperature dependence, protoporphyrin, hematoporphyrin

ABSTRACT: The article describes a study of the EPR signals of chlorophyll and some of its structurally different analogs, namely, pheophytin, ethyl chlorophyllide, hemato- and protoporphyrin, and phthalocyanin and Mg-phthalocyanin. The pigments were examined in the form of solid crystalline samples in glass ampuls, either evacuated or in the presence of air. All the pigments gave a similar

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EPR signal in the dark; the signal was a singlet with a g-factor close to that of a free electron; different pigments displayed small variations in signal width. This observation led to the conclusion that the presence of the unpaired electrons producing the signal is the result of the system of conjugated double bonds of the porphyrin ring, and not the presence or absence of such structural elements as a phytol group, a cyclopentanone ring, or side radicals. In addition, the effect of light on the EPR signal was studied for all the pigments and the effect of temperature and oxygen for chlorophyll a + b only. It was found that all solid pigments produced an increased signal in vacuum which attained its maximum in about 5-10 min. In air the signal (for chlorophyll a + b) increased more than in vacuum. Experiments with films and solutions of chlorophyll a + b indicated that the degree of the pigment aggregation has a significant effect on the signal. The effect of light on phthalocyanin and Mg-phthalocyanin was somewhat different, resulting in an initial increase, then a subsequent decrease of the signal. The temperature dependence of the chlorophyll a + b signal has a maximum at approximately 40C. The nature of the photoinduced signal was not investigated more closely; it is believed that this signal is caused by unpaired

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ACCESSION NR: AP4042475

electrons which arise as a result of an interaction of the excited molecules of chlorophyll with oxygen molecules. It is concluded that the unpaired electrons are dislocated in the conjugated double-bond system or in "active centers" and defects of the crystal lattice of the pigments. The study is considered qualitative, and an evaluation of the quantum yield of the formation of unpaired electrons is suggested. Orig. art. has: 6 figures.

ASSOCIATION: Institut biologicheskoy fiziki AN SSSR, Moscow (Biophysics Institute, AN SSSR); Institut biokhimii im. A. N. Bakha, AN SSSR, Moscow (Biochemistry Institute, AN SSSR)

SUBMITTED: 10Jun62

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NO REF SOV: 003

other: 001

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Pi-4/Pb-4/Pa-4 AFWL/ASD(a)-5/AFETR/ESD(t) ENT(m)/T/EWA(b) L 17717-65 5/0020/64/158/001/0225/0228 ACCESSION NR . AF4045106 Krasnovakiy As (Corresponding member, AN SSSR); Brin, G.P. AUTHOR: of reduced pyridine nucleotides and N-benzyli-TITLE: Photochemistry nicontinamide AN SSSR. Daklady*, v. 156, no. 1, 1964, 225-228 SOURCE: TOPIC TAGS: pyridine nucleotide, diphosphopyridine nucleotide, cohydrogenase, NAD, VNA, nicotinamide adenine dinucleotide, N-benzylniphotochemistry, photocxidation, photoreduction, electron acceptor, electron donor, chlorophyll, bacteriochlorophyll, bacterioviridin, photosensitization ABSTRACT: A detailed investigation of photooxidation of reduced pyridine nucleotides, which may underlie a number of photobiological processes, has been made based on studies of the following types of reduction and oxidation reactions of pyridine nucleotides in the presence of photosensitizing pigments: 1) pigment-photosensitized reduction of NAD (nicotinamide adenine dinucleotide) and BNA (N-benzylni-Card 1/3

L 17717-65 ACCESSION NR: AP4045106

cotinamide) in the presence of hydrogen donors (e.g., ascorbic acid); 2) pigment-photosensitized oxidation of NAD-H2 (reduced form of NAD in the presence of electron acceptors (e.g., 02 and safranine); 3) direct photooxidation of NAD. H2 and BNA. H2 by light excitation at the 365-m wavelength in the presence of electron acceptors (e.g., chlorophyll, safranine T, and riboflavine). The photocxidation of BNA· h2 and NAD. H2 was conducted under conditions of evacuation in a pyridine, alcohol, and aqueous solution, using the 365-mu mercury line for the photoexitation of nucleotides. Atmospheric oxygen, acetaldehyde, NaNO2, and NaNO3 were used as electron acceptors; safranine, hematoporphyrin, and methyl red dyes were used along with chlorophyll, bacteriochlorophyll and bacterioviridin photosynthetic pigments. Results showed that ultraviolet light-activated molecules of NAD-H2 are capable of rapid oxidation when reacting with various electron acceptors. In other experiments, visible-light activation of pigments was studied using NAD+H2 and BNA+H2 exidized by atmospheric oxygen (or safiranine in the presence of chlorophyll). The plant pigments functioned as photosensitizers. The results indicated that vigorous oxidation of pyridine nucleatides and destructive exidation of the pigment

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takes place in visible light. It was observed that bacterial pigments are less stable than chlorophyll, that NAD-H, and BNA-H, form some rather long-lived (more than 60 min) photoexcited species, and that measurable photopotentials arise in NAD-H, and BNA-H, solutions. It was assumed in the discussion of the reaction mechanisms that the formation of a complex with semiquinone structure takes place in pyridine solutions of the investigated nucleotides. Furthermore, the studied reactions were assumed to be connected with the effect of light on enzymatic processes, the photoexidation of pyridine nucleotides, in vivo, and UV caused damage to living cells.

ASSOCIATION: Institut biokhimii im. A., N. Bakh Akademii nauk SSSR (Institute of Biochemistry, Academy of Sciences, SSSR)

SUBHITTED: 28May64

ENCL: 00

SUB CODE: LS

NO REF SOV: '008

OTHER: OOL

Card 3/3

S/0020/64/154/002/0460/0462

AUTHORS: Litvin, F. F.; Sineshchekov, V. A.; Krasnovskiy, A. A. (Corresponding member)

TITLE: On long-wave forms of chlorophyll in photosynthesizing organisms and aggregate structures

SOURCE: AN SSSR. Doklady*, v. 154, no. 2, 1964, 460-462

TOPIC TAGS: long wave spectrum, chlorophyll spectrum, photosynthesis, photosynthesizing organisms, aggregate chlorophyll structure, chlorophyll structure, low temperature spectroscopy, luminescence spectroscopy

ABSTRACT: In the search for a model system closely approximating in vivo conditions for studying spectrum-luminescent properties of natural forms of chlorophyll at -1960, chlorophyll films containing a certain quantity of solvent (ether) were used, i.e. a system ranging from concentrated solution to crystalline pigment layer. Spectrophotometric determinations were conducted on these as well as on chlorophyll-adsorbed chromatographic paper. Five maxima were

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found between 680 and 825 mµ, the first value corresponding to films saturated with solvent, the last to compact films. Shortwave intensity yielded in the same order to long-wave intensity. Comparison with maxima obtained earlier under these conditions from chlorophyll in photosynthesizing organisms showed close_y approximating values. These maxima may correspond to various aggregate chlorophyll forms. The possible composition of these forms is discussed (702-705 mµ may correspond to the "oriented chlorophyll"). Orig. art. has: 3 Figures and 1 Table.

ASSOCIATION: Moskovskiy gosudarstvenny*y universitet im. M. V. Lomonosova (Moscow State University); Institut blokhimii im. A. N. Bakha Akademii nauk SSSR (Institute of Biochemistry, Academy of Sciences, SSSR)

SUBMITTED: 06Sep63

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CIA-RDP86-00513R0008262100

s/0020/64/155/003/0691/0693

Krasnovskiy, A. A. (Corresponding member, AN USSR); Umrikhina, ACCESSION NR: AUTHOR:

On the abiogenic formation of porphine and the participation A. V. of the latter in processes of photochemical electron transfer

SOURCE: AN SSSR. Doklady*, v. 155, no. 3, 1964, 691-693

TOPIC TAGS: Abiogenic synthesis, life origin, porphine, porphine compounds, energy transfer, electron transfer, photosensitization, porphine synthesis

ABSTRACT: This study was prompted by the A. I. Oparin's concept of the formation of primary organic substances on Earth from simple inorganic compounds (abiogenic synthesis) under the influence of UV solar radiation, ionizing radiation, or glow discharge. The study consisted in tracing; the fluorescence of porphine formed from pyrrol and formaldehyde in scaled glass or quartz ampuls. For evacuated ampuls, 8-10-hr heating in a steam bath was required to obtain the distinct rad fluorescence of porphine, while in the presence of air

oxygen a considerable amount, of porphine was formed within an hour. The presence of water or some oxides, such as oxides of silicon, aluminum, titanium, zinc etc., seemed to promote the formation of porphine in the above described reaction. UV radiation was found to transfer from ascorbic acid in the presence of ferrous salts to the transfer from acceptor, which was methyl red in the experiments described. The results confirm the possibility of the abiogenic synthesis of porphine, which could function as a photosensitizer in photochemical lt is noted that ferrous salts probably occurred very widely in the primordial ocean. Gratitude is expressed to Academician A. I. Oparin Institute of Biochemistry, AN SSSR.

ASSOCIATION: Institut biokhimii im. A. N. Bakha Akademii nauk SSSR (Institute of Biochemistry, Academy of Sciences SSSR)

SUBMITTED: 24Dec63

DATE ACQ: 17Apr64

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MOIREF SOV: 005

OTHER: 001

\$/0020/64/155/004/0904/0907

AUTHOR: Umrikhina, A. V. (Corr. member, ANSSSR); Krasnovskiy, A. A. (Corr. member, ANSSSR)

TITLE: Initiation of methyl methacrylate polymerisation by reduced forms of chlorophyll and hematoporphyrin

SOURCE: AN SSSR. Doklady*, v. 155, no. 4, 1964, 904-907

TOPIC TAGS: chlorophyll, hematoporphyrin, tetrapyrrol pigment, reduced chlorophyll, reduced hematoporphyrin, photoreduction, zinc amalgam raduction, chlorophyll ion radical, hematoporphyrin ion radical, free radical, alectron donor, polymerization initiation, methyl methacrylate, chlorophyll analog

ABSTRACT: An attempt was made at the Institute of Biochemistry, Academy of Sciences SSSR, to verify the formation of free radicals in reduction reaction of chlorophyll and its analogs with zinc. Zinc was selected as the reducing agent in order to exclude the presence of electron donor radicals from the liquid phase of the system. The reduction was conducted in Tunberg tubes[sketch shown in the article],

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ACCESSION MR: AP4030791

in the upper part of which chlorophyll (or hematoporphyrin) was reduced in the dark in pyridine solution with sinc amalgam. The reduction was carried out either in the presence of a small amount of glacial acetic acid or without it. The reduced pigment was transferred into the lower part of the tube for spectroscopic study or for chemical tests for the presence of free radicals. The spectral investigation indicated that chlorophyll is reduced to a compound with an absorption peak of 520 m, either with or without glacial acetic acid. Hematoporphyrin without the acid was reduced to a form with an absorption peak of 460 m, and with the acid, to a form with a 670-680-m peak. The reduced forms re very unstable. The reduced chlorophyll form reverts to the initial form within 5-6 min in vacuum; conversion takes place instantaneously on the admission of air into the tube. The hematoporphyrin product is somewhat stabler, but the same conversions also take place. The difference between two reduced forms of the pirments - the photoreduced (e.g., with ascorbic acid) and zinc amalgam-reduced in the dark - is pointed out. The photoreducad form of chlorophyll shows no EPR signal, does not initiate polymerication, and is stable, although its spectrum is similar to that

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of the unstable and vary reactive zinc amalgam-reduced form. It is assumed that the photoreduced form is not afree radical, but a product with saturated valences. The presence of free radicals in the zinc amalgam-reduced form was verified by the initiation of the polymer-ization of methyl methacrylate in the same Tunberg tubes. No polymer-ization took place in the control tests without pigment. The admission of air to the reduced pigment either inhibited the polymerization entirely (as in the case of chlorophyll without acetic acid) or considerably reduced it. On the basis of the above data, the formation of ion-radicals of the pigments by electron transfer from zinc is assumed. The spectral similarity of the photoreduced and zinc amalgam-reduced (in the dark) forms should be taken into account in spectroscopic studies. Orig. art. has: three figures and I table.

ASSOCIATION: Institut biokhimii im. A. N. Bakha Akademii nauk SSSR (Institute of Biochemistry, Academy of Sciences SSSR)

SUBMITTED: 24Dec63

DATE ACQ: 30Apr64 -

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Card 3/3

S/0020/64/156/006/1451/1454

AUTHOR: Rikhireva, G. T.; Krasnovskiy, A. A. (Corresponding member, AN SSSR)

TITLE: Relationship between the chlorophyll state and EPR spectra of plant leaves

SOURCE: AN SSSR. Doklady*, v. 156, no. 6, 1964, 1451-1454

TOPIC TAGS: chlorophyll aggregation state, chlorophyll EPR, photosynthesis, plant pigments, luminescence spectra, chlorophyll luminescence spectra, shortwave chlorophyll, longwave chlorophyll, etiolated plants, chlorophyll formation

ABSTRACT: The present study was undertaken to establish whether variations in the aggregation state of chlorophyll are reflected in the EPR spectra and to contribute to the knowledge of the participation of various components of the plant pigment system in photosynthesis. Two types of experiments were performed: 1) the EPR spectra of greening etiolated leaves were studied, and 2) the same spectra in leaves whose chlorophyll had been changed (e.g., by

heating or swelling in a solvent such as pyridine) were investigated. The EPR and luminescence spectra obtained were also compared. The experimental data indicate that a close relationship exists between the state of chlorophyll, the type of corresponding EPR signal, and the kinetics of the change in this signal, both under illumination and in the dark. Comparison of the EPR and luminescence spectra indicates that the so-called "shortwave" form of chlorophyll is predominant in the etiolated greening leaves, while the "longwave" form accumulates in the green leaves, producing a type of kinetics of the signal change which is characteristic of the green plant. The study was conducted at the Institute of Biophysics and the institute of Biochemistry im. A. N. Bakh, Academy of Sciences SSSR. Orig. art. has: 3 figures.

ASSOCIATION: Institut biologicheskoy fiziki Akademii nauk SSSR (Institute of biological Physics, Academy of Sciences, SSSR); Institut biokhimii im. A. N. Bakha Akademii nauk SSSR (Institute of Biochemistry, Academy of Sciences, SSSR)

SUBHITTED: 14Mar64

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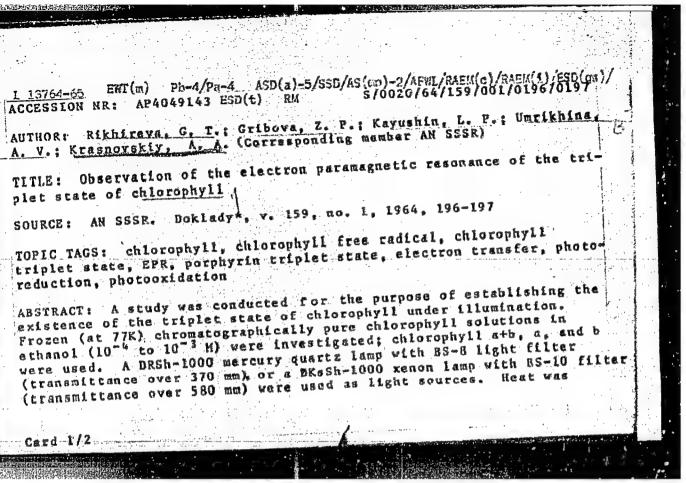
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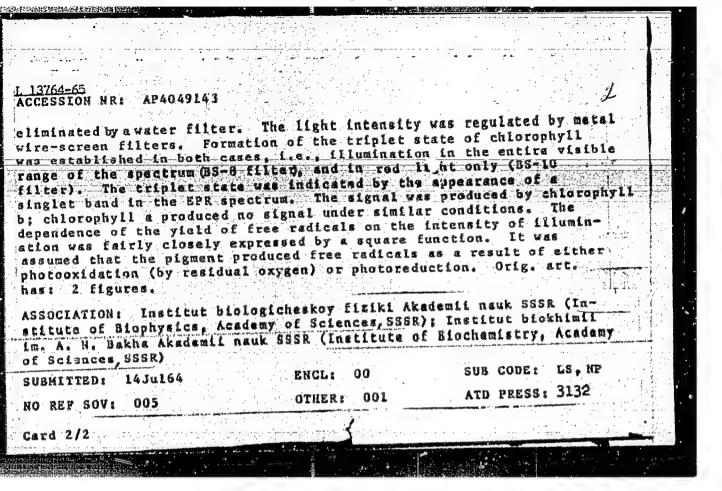
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EMG(j)/EMG(r)/EMT(1)/FS(v)-3/EMG(v)/EMG(a)/EMG(c) Pe-5 DD L 33235-65 ACCESSION NR: AP5002265 8/0218/64/029/006/1132/1142 AUTHOR: Pakshina, Ye. V.; Krasnovskiy, A. A. Investigation of the pheophytinization of chlorophyll bacteriochlorophyll, bacterioviridine, and protochlorophyll, and the effect of light on this reaction SOURCE: Biokhimiya, v. 29, no. 6, 1964, 1132-1142 TOPIC TAGS: plant pigment, pheophytin, chlorophyll, chlorophyll analog, light effect, pheophytinization, photopheophytinization, pH ABSTRACT: Pheophytinization rates of chlorophyll and its analogs were investigated in aqueous alcohol, aqueous acetone, and aqueous pyridine solutions with different pH values produced by the addition of exalic acid or hydrochloric acid. Pheophytinization rates were determined by absorption spectra with SF-5 and SF-10 spectrophotometers. Light effect on the pheophytinization reaction was investigated in a vacuum in Tunberg tubes to avoid photooxidation reactions. A lamp with a condenger and a KS-11 red light filter producing an intensity of 3 x 105 ergs/om2.sec was used. Findings Card 1/3

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show that, on the basis of the pheophytinization reaction rates found for chlorophyll and its analogs, the pigments can be arranged in the following decreasing order: bacterioviridine, chlorophyll a, protochlorophyll and bacteriochlorophyll, and chlorophyll b. The reaction rate is not related to reduction of the "semi-isolated" double bonds of the pigment molecule, but is significantly affected by the electronegative substitutes of the pyrrole nuclei in the molecule. The rapid pheophytinization rate of bacterioviridine, in whose molecule the cyclopentane ring is modified and the carbomethoxy group in the C10 position is absent, attests to the stabilizing effect of this redical on the magnesium bond in the pigment molecule. A comparison of the pheophytinization rates for chlorophylls a and b and the chlorophyllides obtained from them shows that the presence or absence of phytol does not significantly affect the strength of the magnesium bond. Light accelerated the pheophytinization of chlorophylls a, b and protochlorophyll in aqueous pyricine solutions. Possibly the mechanism of "photopheophytinization" differs from that of "dark" pheophytinization. Compounds which inhibit the photoreduction of chlorophyll (carotene, naphthacene, methyl red, safranine, and riboflavin) also inhibit photopheophytinization. Photopheophy-

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	tinization may be rechlorophyll. Orig.	lated to the intermediate pho art. has: 4 figures and 2 to	toreduction of bles.						
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	Submitted: 14May64	ENCL: 00	SUB CODE: LS						
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ENT(m)/EFF(c)/EFF/ENP(1)/T/ENA(c) Pc-lu/Pr-lu/Ps-lu/P1-lu RPL HOWEL/WW/RIM ACCESSION NR: APSOLO171 UR/0020/65/161/902/0399/0402 AUTHOR: Khutaraya, G. V.; Brin. G. P.; Davydov, B. E.; Krentsel, B. A. Krasnovskiy, A. A. (Corresponding member AR SSSR) TITLE: Photosensitizing properties of polyconjugated organic polymers SOURCE: AN SSSR. Doklady, v. 161, no. 2, 1965, 399 402 TOPIC TAGS: photosensitization, conjugated double bond system, polyconjugated polymer, ascorbic acid, oxidation, polyacronitrile, Schiff's base, polymitrile, polyquinoline ABSTRACT: This study investigates the photosensitizing effect of polymers with a system of conjugated double bonds on the oxidation of ascorbic soid. The study was prompted by the fact that photosensitization was established for some crystalline organic dyes and phthalocyanines (semiconfucting substances with conjugated bonds). The Warburg-Barcroft micromanometric method was applied to trace the kinetics of the reaction. The reaction was conducted in aqueous ascorbic acid solution in the prosence of finely powdered polymers under red light (wavelingth, more than 600 mm), white light of an incandescent bulb, or UV light (mercury 365-mu band). The following polymers were used: thermally treated polyacronitrile, heat-polymerised

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ACCESSION NR: APSOLO171

quinoline, polypropiolic acid, polymeric Schiff's bases, polyazines and polynitriles All of these polymers were insoluble, colored solids with absorption maxima in UV and were p-type semiconductors in the air. They could be divided into two groups with respect to their catalytic effect on the oxidation of ascorbic acid: 1) photosensitizing polymers, such as heat-treated polyacronitrile, polyquinolines, polypropiolic acid, and poly-Schiff's bases, all of which promoted the phot-exidation of ascorbic acid, which is not oxidized without catalyst; 2) catalysts in the dark: polynitriles and paracyanogen, the catalytic effect of which in general was inhibited by illumination in the sequence UV light > white light > red light. The following observations were made on the photosensitizing effect of the heat-treated polyacronitrile: 1) the presence of carbonized structures was not essential for the effect, since one of the most carbonized specimens displayed a very weak photosensitizing effect in the UV light and none under the white light; 2) the presence or absence of cross links was of no special influence, since polyacronitrile, heat treated in solution (which precluded the formation of cross links), displayed a rather high photosensitizing effect. Further investigation of the mechanism of the effect is being continued. Orig. art. has: 2 figures and 4 tables. [NC]

ASSOCIATION: Institut neftekhimicheskogo sfames im. A. V. Topchiyeva (Institute' of Petrochemical Synthesis); Institut biokhimii im. A. N. Bakha Akademii nauk (In-Card 2/3

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EWG(1)/EWG(r)/EWT(1)/FS(r)-3/EWG(r)/EWG(a)-2/EWG(c) Po-5 L 41501-65 \$/0020/64/158/003/0730/0733 ACCESSION NRE AP4046386 20 AUTHOR: Krasnovskiv. A. A. (Corresponding Drozdova, H. H. TITLE: Reversible photocheuical exidation and reduction of chlores phyll, bacteriochlorophyll, and bacterioviridic in a viscous mediuni differential absorption spectra of the intermediate form Doklady*, v. 158, no. 3, 1964, 730-733 SOURCE: AN SSSR. TOPIC TAGS: photosynthesis, photoreduction, photooxidation, plant pigments, chlorophyll, bacterichlorophyll, bacterioviridin, absorption spectrum, differential absorption spectrum, chlorophyll intermediate forms ABSTRACT: Techniques for measuring the differential absorption spectra of plant pigments intrumental in photosynthesis were applied for possible identification of the intermediate products in photoexidation or photoreduction. The measurements were conducted in viscous media (alcohol-glycarol mixtures for photoexidation and pyridine-glycarol mixtures for photoreduction) and, in order to stabilise the inter-Card 1/3

L 41501-65 ACCESSION NR: AP4046386

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mediate reaction products, at low temperatures. Chromatographically purified chlorophyll a and b, becteriochlorophyll, bacterioviridin, protochlorophyll, and pheophytin were studied. Oxidation was carried out by a quinone or by oxygen, and reduction, by ascorbic acid or, in the case of bacteriochlorophyll, by Na28 aqueous solution. The oxidative effect of several quinones was tested with chlorophyll s, and p-benzoquinone was selected as producing the bast results, i.e., completeness and 90% reversibility of the reaction. Oxygen in contrast, produced an almost irreversible oxidation. In the process of oxidation, optical density was measured before illumination, at the end of a two-minute illumination period, and in the dark; measurements were made at -70C. The visible range from 400 mu to 750 mu was used for all the pigments except bacteriochlorophyll, for which the IR region to 900 mm was used in addition to the visible range. A similar arrangement for measurements was used for photoreduction. For all the pigments, the differential absorption spectra i.e., the spectra before illumination minus the spectra after illumination (the so-called "dark" spactra), were constructed from the measured optical-density data. It was found that these spectra are essentially similar for

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both photooxidation and photoreduction; a drop in the "red" and "blue" maxima and the appearance of new absorption in the medium range of the visible spectrum were noted, especially in the case of chlorophyll a and bacterioviridin together with certain other minor changes. In the case of bacteriochlorophyll the differences are more pronounced. In all cases the oxidation with quinone and oxygen produced very similar pictures, which fact indicates that actual oxidation takes place in the photoreaction. The similarity between the oxidized and reduced to ms of the pigments indicates that a disruption in the cyclic conjugated system of double bonds takes place. Measurements of model systems, which contain electron donors or acceptors contribute to the identification of the changed chlorophyll forms in vivo. Orig. art. has: 3 figures and 2 tables.

ASSOCIATION: Institut biokhimii im. A.N.Bakha Akademii nauk SSSR (Institute of Biochemistry, Academy of Sciences SSSR)

SUBHITTED: 16 Jun 64

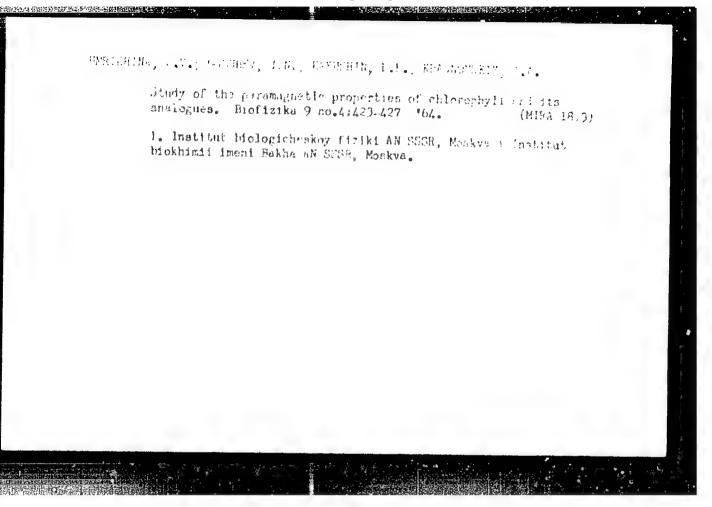
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L 27085-66 ENT(1) SCTB DD	THE T
ACC NRI AP6017429 SOURCE CODE: UR/0217/65/010/002/0242/0245	•
AUTHOR: Karapetyan, N. V.; Krasnovskiy, A. A. ORG: Institute of Biochemistry im. A. N. Bakh, AN SSSR, Moscow (Institut biokhimii AN SSSR)	
TITLE: Changes in fluorescence during the measurement of differential spectra of green photosynthesizing bacteria 2 SOURCE: Biofizika, v. 10, no. 2, 1965, 242-245 TOPIC TAGS: bacteria, photosynthesis, photomultiplier, bacteriology	
AESTRACT: In the determination of differential spectra in the 680-760 millimioron range of the green photosynthesizing bacteria Chlorosaudemonas ethylicum, which contain bacterioviridin, apparent maxima of transmission of light through bacterial suspensions were observed at 700, 725, and 750 millimicrons. The effect of increased transmission was not observed when monochromatic light with these wavelengths was passed through the suspension, while filters transmitting only this light were placed between the suspension and the photomultiplier. This indicated that the effect was not due to discoloration and increased transmission of bacterioviridin at the three maxima. Apparently changes in the fluorescence of bacterioviridin under the effect of the light used in the determination of the spectrum had taken place. Determination, on bacterial suspensions of the spectrum, of excitation of fluorescence of bacterioviridin using monochromatic light indicated the presence of maxima at 680, 700 and 745 millimicrons. The bacteria cultures were given by Ye. N. Kondrat'yeva (Department of Microbiology MBU). Orig. art. has: 2 figures. [JPRS]	1.5
SUB CODE: 06 / SUBM DATE: 29Sep64 / ORIG REF: 004 / OTH REF: 004 Card 1/1 H/ UDC: 577.37	

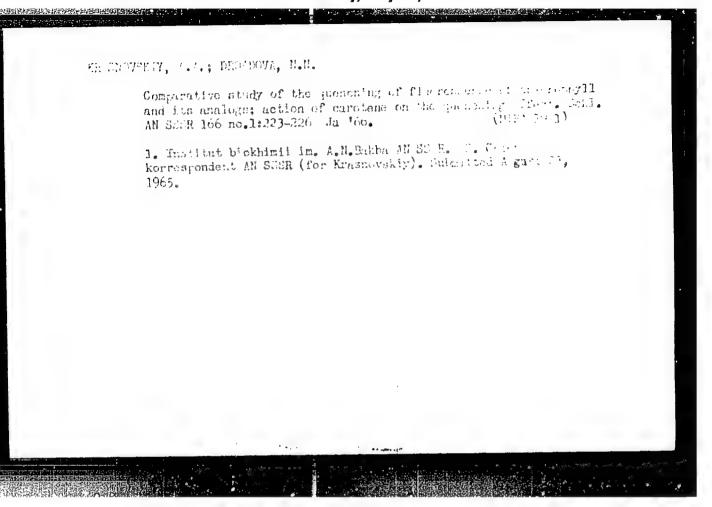
PAKSHINA, Ye.V.; KRASNOVSKIY, A.A.

Study of the pheophytinization of chlorophyll, bacteriochlorophyll, bacterioviridin, protochlorophyll and the effect of light on this reaction. Biokhimia 29 no.6:1132-1142 N-D **164.

(MIRA 18:12)

1. Institut biokhimii imeni A.N.Bakha AN SSER, Moskva. Submitted Mny 14, 1964.

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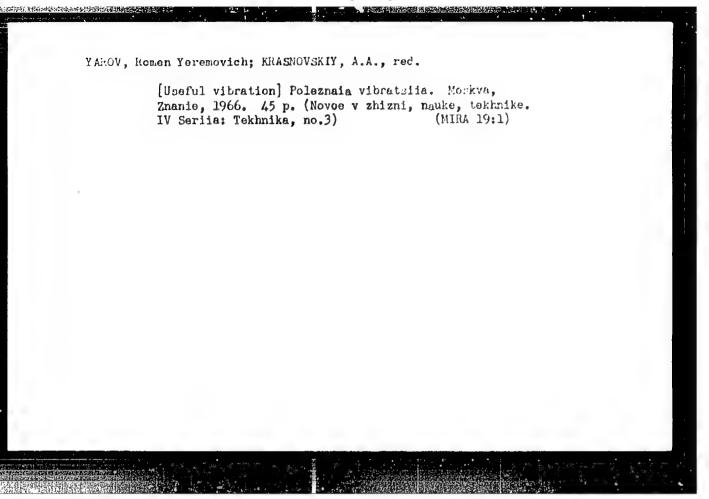
KRASHOVSKIY, A.A., otv. red.; GODNEV, T.M., akademik, red.; MICHIFORCVICH, A,A., prof., red.; TERREIH, A.N., akademik, red.; LITVIN, F.F., red.

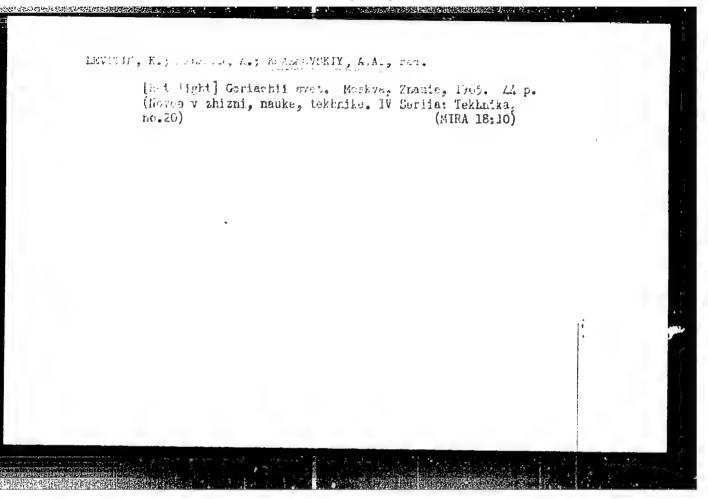
[Biochemistry and biophysics of photosynthesis] Biokhimia i biofizika fotosinteza. Moskva, Nauka, 1965. 318 p.

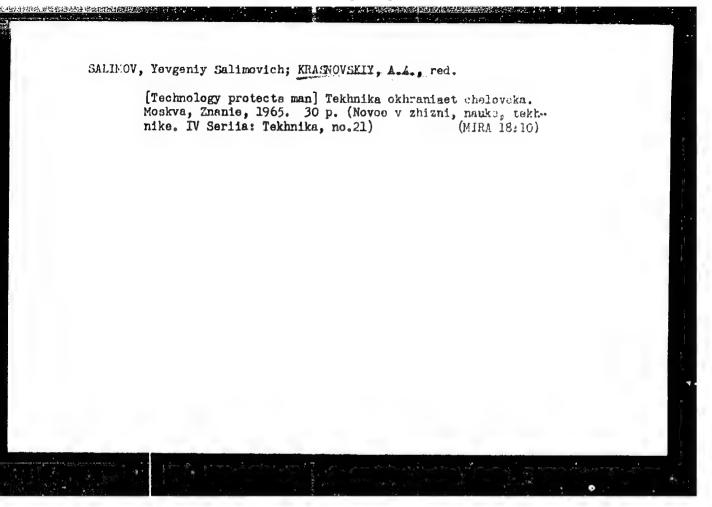
(MIRA 18:10)

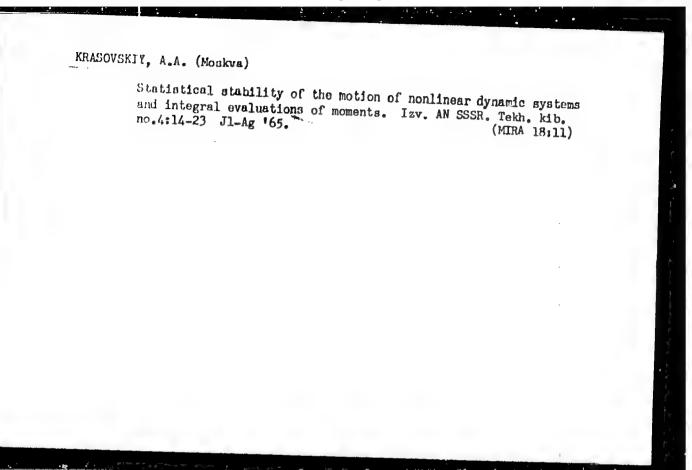
1. Akademiya nauk SSSR. Nauchnyy sovet po probleme "Fotosintez."

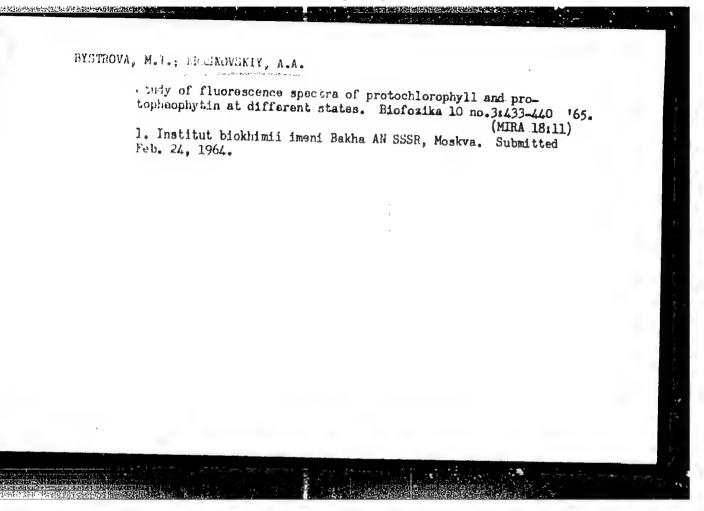
2. Chlen-korrespondent AN SSSR (for Krasnovskiy).

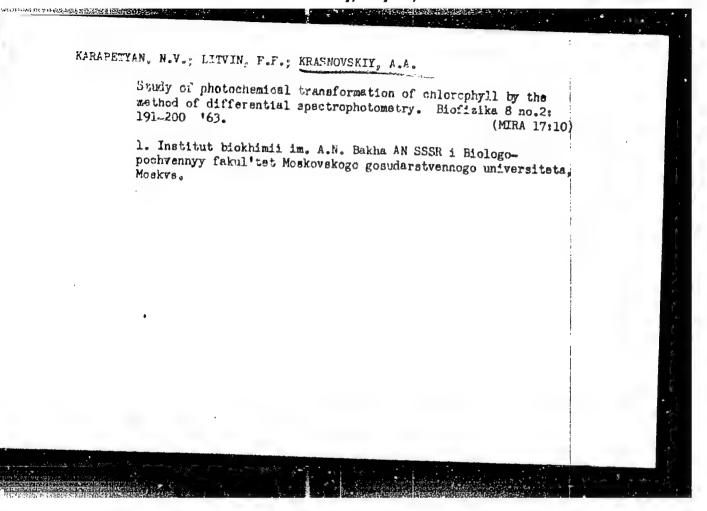












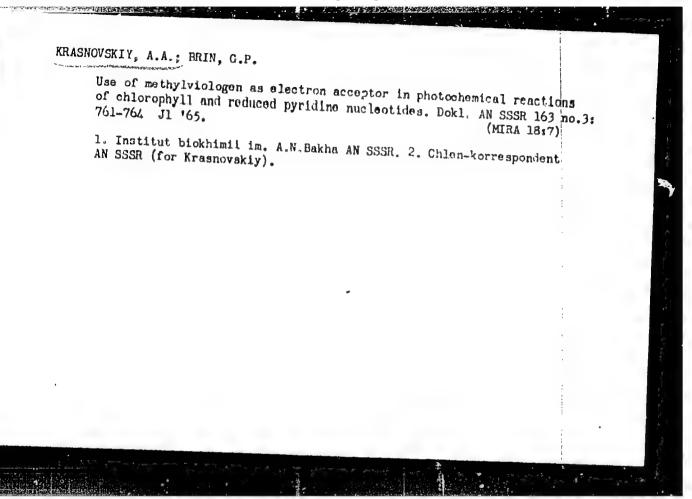
RIKHIREVA, G.T.; UMRIKHINA, A.V.; KAYUSHIN, L.P.; KRASNOVSKIY, A.A.

Formation of triplet and radical states of porphyrin and its derivatives.

Dokl. AN SSSR 163 no.2:491-494 Jl '65. (MIRA 18:7)

1. Institut biologichertoy fiziki AN SSSR i Institut biokhimii im.

A.N.Bakha AN SSSR. 2. Chlen-korrespondent AN SSSR (for Krasnovskiy).



KARAFETYAN, N.V.; KRASHOVSKIY, A.A.

Changes in fluorescence during the measurement of differential spectra of green photosynthetizing bacteria. Biofizika 10 no.28242-245 65.

(MIRA 18:7)

1. Institut bickhimii imeni Bakha AN SSSR, Moskva.

KHUTAREVA, G.V.; BRIN, G.P.; DAVYDOV, B.E.; KRASNOVSKIY, A.A.; KHENTSEL*, B.A.

Photosensitizing properties of polyconjugate organic polymers. Dokl. AN SSSR 161 no.2:399-402 Mr 165.

1. Institut neftekhimicheskogo sinteza im. A.V. Topchiyeva i Institut blokhimii im. A.N. Rakha AN SSER. 2. Chlen-korrespondent AN SSSR (for Krasnovskiy).

RIKHIREVA, G.T., GRIPOVA, Z.P., KAYUSHIN, L.P., UMRIKHINA, A.V., KRASNOVSKIY, A.A.

Observation of the electron paramegnetic resonance in the triplet state of chlorophyll. Pokl, AN SSSR 159 no.1:196-197 N 164. (MIRA 17:12)

1. Institut biologicheskoy firtki AN SSSR i lnabitut biokhimii im. A.N. Bakha AN SSSR. 2. Chlem-korrespondent AN SSSR (for Krasnovskiy).

EWT(m)/EWP(t)/ETI L 42180-66 IJP(c) JD ACC NRI UR/0020/66/168/005/1100/1103 AP6021609 SOURCE CODE: Krasnovskiy, A. A. (Corresponding member AN SSSR); Brin, G. P. AUTHOR: ORG: Institute of Biochemistry im. A. N. Bakh, Academy of Sciences, SSSR (Institut: biokhimii Akademii nauk SSSR) TITIE: Photochemical liberation of oxygen in aqueous solutions of ferric iron; sensitization with tungsten, titanium and zinc oxides SOURCE: AN SSSR. Doklady, v. 168, no. 5, 1966, 1100-1103 photochemistry, oxygen, zinc oxide, titanium dioxide, tungsten compound, 41820, Aqueous Solurion 1.7 TOPIC TAGS: QUANTUM ABSTRACT: The quantum yields of liberation of oxygen in the photochemical oxidation of water (containing potassium ferricyanide or ferric ammonium alum) and the stoichiometry of the reactions taking place were studied in the presence of the photocatalysts ZnO, TiO2, and WO3. The sensitized liberation of oxygen was measured in glass vessels, i.e. under conditions where the direct photochemical liberation of oxygen is practically absent. The 365 mm line of mercury was used as the exciting light in experiments with 2n0 and $7i0_2$; the 365, 404, and 436 mm linus were used for 80_3 . The quantum yield of the reactions was found to be of the order of 10^{-2} ; in nitrogen, the quantum yield is higher than in air. The highest yield of liberation of oxygen was obtained in the sys; tem containing WO3 and ferric ammonium alum. Although the efficiency of the sensitized reactions is low, it is much higher than that of direct photoreactions (without photo-UDC: 541.1+541.8+541.4 Card

CC NR: AP6021	ng place in these syst	ems in shorter UV on	light in quartz ve	ssels. Possi-
	$F_{\theta_2}(SQ_4)_3 + H_2O \rightarrow$	2FeSQ ₄ + 1/2 02 +	H ₂ SQ ₊	:
14 counced	Onin art. hasi 2 f	igures and 1 table	•	
re discussed.	SUBM DATE: 20Dec65/	ORIG REF: 007/	OTH REF: 007	
OB CODS: OIL				
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14				

ACC NR: AP7002393

SOURCE CODE: UR/0020/66/171/005/1201/1204

AUTHOR: Karapetyan, N. V.; Krakhmaleva, I. N.; Krasnovskiy, A. A. (Corresponding member AN SSSR)

ORG: Institute of Biochemistry im. A. N. Bakh, Academy of Sciences SSSR (Institut biokhimii Akademii nauk SSSR)

TITLE: Effect of heat inactivation on differential absorption spectra of purple photosynthesizing bacteria

SOURCE: AN SSSR. Doklady, v. 171, no. 5, 1966, 1201-1204

TOPIC TAGS: bacteria, chlorophyll, temperature dependence, absorption spectrum, fluorescence spectrum

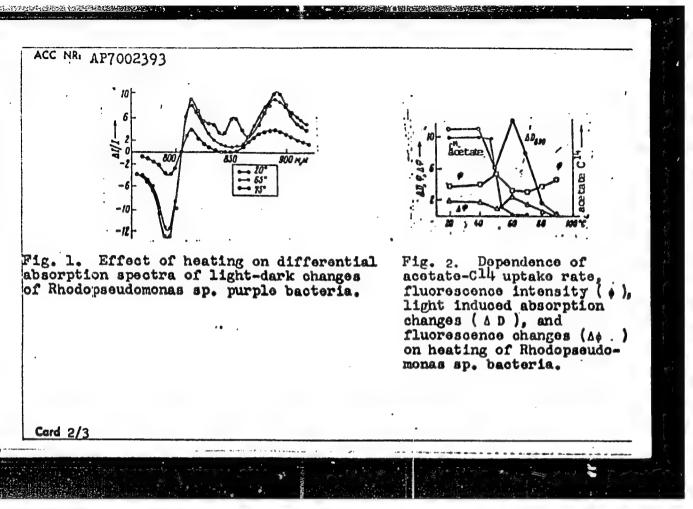
ABSTRACT: In experiments on Rhodopseudomonas sp. and Chromatium purple bacteria, the effect of heat inactivation on bacteriochlorophyll was determined by differential absorption spectra, photosynthesis rate. (acetate-Cl4 uptake) and absorption and fluorescence spectra. Bacterial suspensions in a culture medium were heated to temperatures of 40 to 90c and then were subjected to freezing (-196c) and thawing. Differential spectra were measured following a five sec "light period" and a one min "dark period." Absorption spectra were measured with an SF-10 spectrophotometer and fluorescence spectra were measured with a

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spectrofluorimeter assembled by Yu. Ye. Yerokhin in the laboratory. Findings indicate that photosynthetic activity is not affected by heating at 50c, but drops sharply at 55c and is completely depressed at 50c. At 60c the absorption and fluorescence spectra are not affected, but at 70c a change in the bacteriochlorophyll takes place as expressed by reduced absorption at 890 mr and increased fluorescence at 910 mr. Heating to 80c and higher produces significant absorption and fluorescence spectra changes. The differential spectra (see Fig. 1) show that the dependence of absorption changes at 790, 810, 850 and 890 mr is complex. A possible explanation is offered for the high sensitivity of photosynthesis to heat. Heat inactivates some of the photosynthetic enzyme reactions and this leads to an accumulation of photochemically changed pigment molecules due to blocking of electron transfer chains, which in turn leads to a gradual breakdown of the pigment-protein complex and photoreactivity. The authors express their thanks to Ye. N. Kondrat'yeva for supplying the bacterial cultures. Orig. art. has: 4 figures.

SUB CODE: 06/ SUBM DATE: 29Aug66/ ORIG REF: 008/ OTH REF: 004

Card 3/3

KRASNOVSKIY, A.I.; KOBOSNIDZE, G.I.; VAGRAMYAN, A.T.

Role of passivation in the electrodeposition of nickel. Zhur. fiz. khim. 36 no.4:714-719 Ap '62. (MIRA 15:6)

1. Akademiya nauk SSSR, Institut fizicheskoy khimii. (Passivation) (Nickel plating)

RRASHMISKY, AZ.

TREASURE ISLAND BIBLIOGRAPHICAL REPORT · PHASE I

AID 358 - I

BOOK

Call No.: TN672.V8

Author: KRASNOVSKIY, Eng.

TEMPERING OF FILES BY HIGH FREQUENCY CURRENT AT MIASS Full Title:

FILE MANUFACTURING PLANT

Transliterated Title: Zakalka napil'nikov tokami vysokov chastoty

na Miasakom napilcchnom zavode

Publishing Data

All-Union Scientific Engineering and Technical Originating Agency:

Society of Machine Euilders. Urals Branch

State Scientific and Technical Publishing House of Machine Building Literature ("Mashgiz") Publishing House:

Date: 1950

No. pp.: 10

No. of copies: 3,000

Text Data

This is an article from the book: VSESOYUZNOYE NAUCHNOYE INZHENERNO-TEKHNICHESKOYE OBSHCHESTVO MASHINOSTROITELEY. URAL'SKOYE OTDELENIYE, THERMAL TREATMENT OF METALS - Symposium of Conference (Termicheskaya obrabotka metallov, materialy konferentsii) (p.303-312), see AID 223-II The surface tempering of files of different sizes and

Coverage:

shapes by high frequency current is described and illustrated with the inductors pertinent to each type or file. Automatic and non-automatic tempering machines are outlined

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CIA-RDP86-00513R0008262100 APPROVED FOR RELEASE: Monday, July 31, 2000

Zakalka napil'nikov tokami vysokoy chastoty na Miasskom napilochnom zavode .ID 358 - I

with the indication of production data and quality of the output. Advantages of tempering by high frequency current are discussed and compared with the results obtained in gas furnaces of the conventional type.

3 drawings and 2 tables. Purpose: For scientific workers

Facilities: None

No. of Russian and Slavic References: None

Available: Library of Congress.

2/2